## **Acid And Bases Ph Phet Lab Answers**

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involved in any changes that take place).

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pH practice - Answers. 1) What is the pH and pOH of a 1.2 x 10-3 HBr solution? pH: 2.9 pOH: 11.1. 2) What is the pH and pOH of a 2.34 x 10-5 NaOH solution? ... Strong acids and bases would be stronger electrolytes because they produce more ions in solution than weak acids and bases...

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