

## Answers For Polyatomic Ions Packet

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Polyatomic Ions Worksheet Polyatomic Practice 1. Name or write the formula for the following polyatomic ions sulfate  $\text{CO}_3^{2-}$  nitrite  $\text{MnO}_3^-$  1-perphosphate  $\text{SO}_5^{2-}$  2-hypiodite  $\text{BrO}_2^-$  1-chlorite  $\text{CO}_4^{2-}$  2-phosphite  $\text{PO}_5^{3-}$  3-percarbonate  $\text{ClO}_1^-$  bromate  $\text{IO}_2^-$  1-hyposulfite  $\text{PO}_4^{3-}$  ...

Common Polyatomic Ions Name(s) Formula Name(s) Formula ammonium  $\text{NH}_4^+$  acetate  $\text{CH}_3\text{COO}^-$   $\text{C}_2\text{H}_3\text{O}_2^-$  bromate  $\text{BrO}_3^-$  carbonate  $\text{CO}_3^{2-}$  chlorate  $\text{ClO}_3^-$  chlorite  $\text{ClO}_2^-$  chromate  $\text{CrO}_4^{2-}$  cyanide  $\text{CN}^-$  dichromate  $\text{Cr}_2\text{O}_7^{2-}$  2-hydrogen carbonate bicarbonate  $\text{HCO}_3^-$  hydrogen sulfate bisulfate  $\text{HSO}_4^-$  hydrogen phosphate biphosphate  $\text{HPO}_4^{2-}$  hydroxide  $\text{OH}^-$

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Polyatomic ions are charged groups of atoms. An example is ammonium ion,  $\text{NH}_4^+$ . It has five atoms (one nitrogen and four hydrogens) that share a charge of +1. The polyatomic ions remain intact, and parentheses may be required when using subscripts. For example, ammonium chloride is  $\text{NH}_4\text{Cl}$

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and ammonium sulfide is  $(\text{NH}_4)_2\text{S}$ . Ammonium is the ...

Polyatomic Ions Worksheet Answer Key Polyatomic Practice 1. Name or write the formula for the following polyatomic ions sulfate  $\text{SO}_4^{2-}$  carbonate  $\text{CO}_3^{2-}$  nitrite  $\text{NO}_2^-$  permanganate  $\text{MnO}_4^-$  perphosphate  $\text{PO}_4^{3-}$  persulfate  $\text{SO}_5^{2-}$  hypoiodite  $\text{IO}^-$  bromite  $\text{BrO}_2^-$  chlorite  $\text{ClO}_2^-$  percarbonate  $\text{C}_2\text{O}_4^{2-}$  phosphite  $\text{PO}_3^{3-}$  perphosphate  $\text{PO}_5^{3-}$  percarbonate  $\text{C}_2\text{O}_4^{2-}$  ...

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Polyatomic Ions Worksheet Polyatomic Practice 1. Name or write the formula for the following polyatomic ions sulfate  $\text{CO}_3^{2-}$  nitrite  $\text{MnO}_3^-$  perphosphate  $\text{SO}_5^{2-}$  hypoiodite  $\text{BrO}_2^-$  chlorite  $\text{CO}_4^{2-}$  phosphite  $\text{PO}_5^{3-}$  percarbonate  $\text{ClO}_1^-$  bromate  $\text{IO}_2^-$  hyposulfite  $\text{PO}_4^{3-}$  ...

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or molecule that has  $\downarrow$ , polyatomic ions are on the

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Common Polyatomic Ions Name(s) Formula Name(s) Formula ammonium  $\text{NH}_4^+$  acetate  $\text{CH}_3\text{COO}^-$   
 $\text{C}_2\text{H}_3\text{O}_2^-$  bromate  $\text{BrO}_3^-$  carbonate  $\text{CO}_3^{2-}$  chlorate  $\text{ClO}_3^-$  chlorite  $\text{ClO}_2^-$  chromate  $\text{CrO}_4^{2-}$  cyanide  $\text{CN}^-$   
dichromate  $\text{Cr}_2\text{O}_7^{2-}$  hydrogen carbonate bicarbonate  $\text{HCO}_3^-$  hydrogen sulfate bisulfate  $\text{HSO}_4^-$   
hydrogen phosphate biphosphate  $\text{HPO}_4^{2-}$  hydroxide  $\text{OH}^-$

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Polyatomic ions are charged groups of atoms. An example is ammonium ion,  $\text{NH}_4^+$ . It has five atoms (one nitrogen and four hydrogens) that share a charge of +1. The polyatomic ions remain intact, and parentheses may be required when using subscripts. For example, ammonium chloride is  $\text{NH}_4\text{Cl}$  and ammonium sulfide is  $(\text{NH}_4)_2\text{S}$ . Ammonium is the ...

Polyatomic Ions Worksheet Answer Key Polyatomic Practice 1. Name or write the formula for the following polyatomic ions sulfate  $\text{SO}_4^{2-}$  carbonate  $\text{CO}_3^{2-}$  nitrite  $\text{NO}_2^-$  permanganate  $\text{MnO}_4^-$  phosphate  $\text{PO}_4^{3-}$  persulfate  $\text{SO}_5^{2-}$  hypoiodite  $\text{IO}^-$  bromite  $\text{BrO}_2^-$  chlorite  $\text{ClO}_2^-$  percarbonate  $\text{C}_2\text{O}_4^{2-}$  phosphite  $\text{PO}_3^{3-}$  perchlorate  $\text{ClO}_4^-$  ...

**Answers For Polyatomic Ions Packet** pdf free pdf download now source 2 which ions in the table are polyatomic ions www answers com categories science chemistry no baso4 is an ionic compound which consists of a polyatomic ion and a monoatomic ion a polyatomic ion is an ion a charged atom

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or molecule that has  $1-$ , polyatomic ions are on the

Polyatomic Ions Worksheet Polyatomic Practice 1. Name or write the formula for the following polyatomic ions sulfate  $\text{CO}_3^{2-}$  nitrite  $\text{MnO}_3^-$  1-perphosphate  $\text{SO}_5$  2-hypiodite  $\text{BrO}_2^-$  1-chlorite  $\text{CO}_4^{2-}$  2-phosphite  $\text{PO}_5^{3-}$  3-percarbonate  $\text{ClO}_1^-$  bromate  $\text{IO}_2^-$  1-hyposulfite  $\text{PO}_4^{3-}$  ...

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POLYATOMIC IONS CHART NAME: 1- Charge 2- Charge 3- Charge Aluminate  $\text{AlO}_2^-$  ?  
Tetraborate  $\text{B}_4\text{O}_7^{2-}$ -Arsenite  $\text{AsO}_3^{3-}$ ? Metaborate  $\text{BO}_2^-$

BONDING NOTE PACKET - 10 7) The sum of the oxidation numbers in polyatomic ions must equal the charge on the ION (see Table E). -2 Ex:  $(\text{Cr}_2\text{O}_7)^{2-}$  Cr: \*Compounds containing POLYATOMIC IONS have BOTH IONIC and COVALENT bonds. Practice: Identify and list the polyatomic ions in the following compounds (use Table E).

Chemistry: Ionic Compounds: Polyatomic Ions with Multiple-Charge Cations. Write the name of each of the following compounds. 1.  $\text{V}(\text{ClO}_3)_5$   $\text{V}^{5+}$  5 . ClO. 3. 1- 1. vanadium (V) chlorate. 2.  $\text{Re}(\text{SO}_4)_3$   $\text{Re}^{6+}$  3 . SO. 4. 2- 2. rhenium (VI) sulfate. 3.  $\text{Os}(\text{IO}_3)_3$   $\text{Os}^{3+}$  3 . IO. 3. 1- ...

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