

## Ap Chemistry Chapter 1 4 Test

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AP Chem: Chapter 4 Practice Multiple Choice Questions Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. What mass of silver nitrate, AgNO<sub>3</sub>, is required to prepare 800. g of 3.50% solution of AgNO<sub>3</sub>? a. 24.6 g b. 26.7 g c. 27.0 g d. 25.5 g e. 28.0 g \_\_\_\_ 2.

24/9/2017 · View Test Prep - Practice test answer key.pdf from CHEM REGULAR at Newton South High School. AP CHEMISTRY CHAPTER 1 AND 2 PRACTICE TEST MULTIPLE CHOICE: Select the one best answer and place it in the

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Modern Chemistry 2 Electron Arrangement Chapter 4 PRE TEST, continued \_\_\_\_\_ 4. The electron configuration below violates a. the Pauli exclusion principle. b. the Aufbau principle. c. Hund's rule. d. Both (a) and (c) \_\_\_\_\_ 5. A photon is emitted from a gaseous atom when an electron moves to its ground state from a(n) a. inner shell.

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1. Enter the Y values in column A. 2. Enter the X values in column B. 3. Press the chart button in the toolbar, OR under Insert in the menu, select Chart. 4. Select plot type "XY scatter". Press . 5. Set the data range by selecting all the data. (Click in a corner and drag the mouse until all boxes are selected. Do NOT include the titles.) 6.

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