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2 + 3 H₂ → 2 NH₃ 2 Cu₂O + C → 4 Cu +
CO₂ 2 H₂O → 2 H₂O + O₂ Hint: Add
the O atoms on the product side together

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2. $\text{Mg} + 2 \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
3. $\text{Cl}_2 + 2 \text{KI} \rightarrow$

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(1) introduce you to the U-ExCEL
Balancing Act Program, (2) offer step by
step instructions to perform and practice a

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(c) $\text{FeBr}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + \text{HBr}$
(d) $\text{C}_4\text{H}_6\text{O}_3 + \text{H}_2\text{O} \rightarrow \text{C}_2\text{H}_4\text{O}_2$ (e) C_2H_4

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+ O₂ ??? CO₂ + H₂O (f) C₄H₁₀O + O₂ ???
CO₂ + H₂O (g) C₇H₁₆ + O₂ ??? CO₂ +
H₂O (h) H₂SiCl₂ + H₂O ??? H₈Si₄O₄ +

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HCl (i) $\text{HSiCl}_3 + \text{H}_2\text{O} ?? \dots$

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$2 + 3 \text{H}_2 + 2 \text{NH}_3 + 2 \text{Cu}_2\text{O} + \text{C} \rightarrow 4 \text{Cu} + \text{CO}_2 + 2 \text{H}_2\text{O}$
 $2 + 2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$ Hint: Add the O atoms on the product side together when doing the counts. Page 2 Practice

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$2 \text{KCl} + \text{I}_2$ 4.

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Problems $2 \text{ Ca} + \text{O}_2 \rightarrow 2 \text{ CaO}$ $\text{N}_2 + 3 \text{ H}_2 \rightarrow 2$

NH_3 $2 \text{ Cu}_2\text{O} + \text{C} \rightarrow 4 \text{ Cu} + \text{CO}_2$ $2 \text{ H}_2\text{O}_2 \rightarrow 2$

$\text{H}_2\text{O} + \text{O}_2$ Hint: Add the O atoms on the

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product side together when doing the counts. Page 2 Practice Problems 1. $2 \text{Na} + \text{MgF}_2 \rightarrow 2 \text{NaF} + \text{Mg}$ 2. $\text{Mg} + 2 \text{HCl} \rightarrow \text{MgCl}_2$

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(c) $\text{FeBr}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + \text{HBr}$

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(d) $C_4H_6O_3 + H_2O \rightarrow C_2H_4O_2$ (e) $C_2H_4 + O_2 \rightarrow CO_2 + H_2O$ (f) $C_4H_{10} + O_2 \rightarrow CO_2 + H_2O$ (g) $C_7H_{16} + O_2 \rightarrow CO_2 + H_2O$ (h) $H_2SiCl_2 + H_2O \rightarrow H_8Si_4O_4 +$

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HCl (i) $\text{HSiCl}_3 + \text{H}_2\text{O} ?? \dots$

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 $2 + 2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O} + \text{O}_2$ Hint: Add the O atoms on the product side together when doing the counts. Page 2 Practice

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NH_3 $2 \text{Cu}_2\text{O} + \text{C} \rightarrow 4 \text{Cu} + \text{CO}_2$ $2 \text{H}_2\text{O}_2 \rightarrow 2$

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Balance each of the following equations. (a)

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(c) FeBr₃ + H₂SO₄ ??? Fe₂(SO₄)₃ + HBr
(d) C₄H₆O₃ + H₂O ??? C₂H₄O₂ (e) C₂H₄
+ O₂ ??? CO₂ + H₂O (f) C₄H₁₀O + O₂ ???

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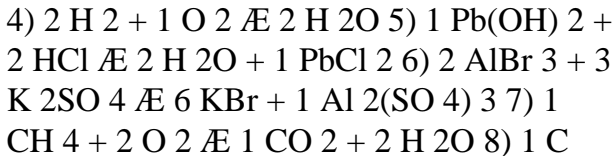
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