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Assessment 36. a. Two formula units  
 $\text{KClO}_3$  decompose to form two formula

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units KCl and three molecules O<sub>2</sub>. b.

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Table 12-1 summarizes the relationships that can be determined from a balanced

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chemical equation. 354 Chapter 12  
Stoichiometry Relationships Derived from a  
Balanced Chemical Equation Iron Oxygen 0  
Iron(III) oxide  $4\text{Fe}(s) + 3\text{O}_2(g) \rightarrow 2\text{Fe}_2\text{O}_3$

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3(s) 4 atoms Fe 3 molecules O 2 0 2  
formula units Fe 2O 3 4 moles Fe 3 moles  
O 2 0 2 moles Fe 2O 3 223.4 g Fe ...

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Table 12-1 summarizes the relationships that can be determined from a balanced chemical equation. 354 Chapter 12  
Stoichiometry Relationships Derived from a

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## Test Review Answers

Balanced Chemical Equation Iron Oxygen  
Iron(III) oxide  $4\text{Fe}(s) + 3\text{O}_2(g) \rightarrow 2\text{Fe}_2\text{O}_3(s)$   
4 atoms Fe 3 molecules  $\text{O}_2$  2 formula units  $\text{Fe}_2\text{O}_3$   
4 moles Fe 3 moles  $\text{O}_2$  2 moles  $\text{Fe}_2\text{O}_3$

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0 2 0 2 moles Fe 2O 3 223.4 g Fe ...

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Assessment 36. a. Two formula units  
 $\text{KClO}_3$  decompose to form two formula

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units KCl and three molecules O<sub>2</sub>. b.

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Defining Stoichiometry pages 368–372

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Practice Problems pages 371–372 1.

Interpret the following balanced chemical equations in terms of particles, moles, and mass. Show that the law of conservation of

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mass is

From the balance above, it is certain that you  
compulsion to approach this **Chapter 12  
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