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**General Science Solutions for
Class 8 Science Chapter 14 ...**

*Let the mass of B M. Let the change
in temperature be T for both the
bodies, A and B. The amount of heat*

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*in a body is given as. $Q = m \times c \times \Delta T$
For body A, $Q = m \times c \times \Delta T$? $\Delta T = \frac{Q}{m c}$ (i) For body B, $4 Q = M \times 2 c \times \Delta T$? $M = \frac{4 Q}{2 c \times \Delta T}$ From (i),
 $\Delta T = \frac{Q}{m c}$? $M = \frac{4 Q}{2 c \times \frac{Q}{m c}} =$*

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2 m.

**Solved: PROJECTS FOR
CHAPTER FOURTEEN 1. Heat**

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Equation ...

*question: projects for chapter
fourteen 1. Heat Equation The
Function $T(x, Y, Z, T)$ Is A Solution
To The Heat Equation And Gives The*

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*Temperature At The Point (x, Y, Z) In
3-space And Time T . The Constant K
Is The Thermal Conductivity Of The
Medium Through Which The Heat Is*

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Flowing.

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Solution, Chapter 14, Problem 6 ...

15/5/2020 · We'll write our kilojoules in units of joules by multiplying by 10 to the 3 since that's what the prefix 'kilo' means and we have a

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formula that the heat added is the mass times its specific heat of the substance times the change in temperature and our job here is to find out what c is and then use that to

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*figure out what material we are
dealing with in this table of specific*

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heats.

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Solution, Chapter 14, Problem 30 ...

Question. (a) Calculate the rate of heat conduction through house walls that are 13.0 cm thick and that have an average thermal conductivity

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*twice that of glass wool. Assume
there are no windows or doors. The
surface area of the walls is 120 m^2*

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and their inside surface is at 18.0 ? ...

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Science Chapter 14 - Sources ...

Answer: Geothermal power plants use heat of the Earth to generate electricity. This heat energy of the Earth is known as geothermal

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energy. When there are geological changes, the molten rocks present in the core of the earth are pushed to the earth's crust. This forms regions

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of hot spot.

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Science Chapter 14 Sources of ...

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Class 10 Science Chapter 14
Textbook Chapter End Questions.
Question 1 A solar water heater*

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*cannot be used to get hot water on
(a) a sunny day (b) a cloudy day (c)
a hot day (d) a windy day Answer:
(b) A cloudy day. Question 2 Which
of the following is not an example of*

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*a biomass energy source ? (a) Wood
(b) Gobar gas (c) Nuclear energy*

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CHAPTER 14

Answers to Chapter Post-Test. If you missed an answer, study the text section and problem-solving skill given in parentheses after the

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answer. 1. d (14.2, PS Sk. 2) 2.

Predominantly reactant; $Q_c = 3.63$ (10^3), hence equilibrium has not been reached. (14.4, 14.5, PS Sk. 4) 3. a.

False. The equilibrium will favor the

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increased formation of H₂(g) ...

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Chapter 14: Heat & Temperature -

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Holt Physical Science ...

Answers: The air is a poor conductor of heat. The pan is a good insulator of heat. The air is a good conductor of heat. The pan is hotter than the

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air. The pan is a poor conductor of heat.

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Chapter 14: Temperature and Heat - Welcome to Laulima!

- *When heat flow is considered, some materials, like metals, are good transmitters of heat energy. We term*

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*these materials to be thermal
conductors. • Materials like
Styrofoam are poor conductors of
heat and will, in fact, severely
restrict heat flow (like the one*

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*described above). Materials that
conduct heat ...*

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**NCERT Solutions for Class 10
Science Chapter 14 - Sources ...**

Answer: Geothermal power plants use heat of the Earth to generate electricity. This heat energy of the

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Earth is known as geothermal energy. When there are geological changes, the molten rocks present in the core of the earth are pushed to the earth's crust. This forms regions

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of hot spot.

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Science Chapter 14 Sources of ...

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**Answer Key Chapter 14 -
University Physics Volume 2 |**

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OpenStax

Check Your Understanding. 14.1.

$4.77 \times 10^{-2} \text{ V}$. $4.77 \times 10^{-2} \text{ V}$.

14.2. a. decreasing; b. increasing;

Since the current flows in the

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opposite direction of the diagram, in order to get a positive emf on the left-hand side of diagram (a), we need to decrease the current to the left, which creates a reinforced emf

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where the positive end is on the ...

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CHAPTER 14

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False. The equilibrium will favor the

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*increased formation of $H_2(g)$ and
 $O_2(g)$.*

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**Solved: Experiment 14 Heat
Effects And Calorimetry At Is A ...**

*Experiment 14 Heat Effects and
Calorimetry at is a form of energy.
sometimes called thermal energy,*

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that can pass spontaneously from an object at high temperature to an object at a lower temperature. If two objects are in contact, they will given sufficient time, both reach the same

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*temperature. Heat flow is ordinarily
measured in a device called a ...*

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Worksheet 14: Practice Exam 3 Answer Key

1. They are all so similar to their atomic forms that the heat of formation is negligible. 2. The heats

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of formation for homonuclear gases are measured at equilibrium, where $\Delta H = 0$. 3. The heat of formation for every element is reported with respect to its homonuclear diatomic

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gaseous form. 4.

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Chapter 14 Intermolecular Forces

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Answers to Practice Problems
Example 1. Polar or Nonpolar?*

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CCl₄ Nonpolar Dispersion (or
London) B. NH₃ Polar H bond C.
HCl Polar Dipole D. OF₂ Polar*

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Dipole Example 2.

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Heat - Welcome to Laulima!

- *When heat flow is considered, some materials, like metals, are good transmitters of heat energy. We term these materials to be thermal*

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conductors. • Materials like Styrofoam are poor conductors of heat and will, in fact, severely restrict heat flow (like the one described above). Materials that

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conduct heat ...

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Science Chapter 14 - Sources ...

Answer: Geothermal power plants use heat of the Earth to generate electricity. This heat energy of the Earth is known as geothermal

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energy. When there are geological changes, the molten rocks present in the core of the earth are pushed to the earth's crust. This forms regions

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of hot spot.

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gaseous form. 4.

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Answers to Practice Problems
Example 1. Polar or Nonpolar?*

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CCl₄ Nonpolar Dispersion (or
London) B. NH₃ Polar H bond C.
HCl Polar Dipole D. OF₂ Polar*

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Dipole Example 2.

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Chemistry, 7e (Wade) Chapter 14
Ethers, Epoxides, and Sulfides 1)*

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*What is the hybridization of the oxygen atom in dialkyl ethers? A) sp^3 B) sp^2 C) sp D) s E) p Answer: A
Diff: 1 Section: 14.2 2) Which of the following corresponds to the COC*

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bond angle in dimethyl ether?

lab14 - Experiment 14 Heat Effects

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and Calorimetry Purpose ...

*Experiment 14: Heat Effects and
Calorimetry Purpose: To fully
understand the effects of heat and to
calculate the specific heat of an*

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*unknown. Experimental Procedure:
Part A: Specific Heat 1. From your
instructor obtain a calorimeter, a
sensitive thermometer, a sample of
metal in a large stoppered test tube.*

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*The calorimeter consists of two
nested expanded polystyrene coffee*

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cups fitted with a ...

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Answers

*Chapter 14 and 15 Study Guide
Answers. Section 14-1.*

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Biogenesis is the principle that all

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living things. come from other living things. 2. Spontaneous generation is the supposed origin of. living things

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from nonliving things.

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Chapter 14: Organic Chemistry

14.1: Bonding in Organic

Compounds (2) 14.2: Aromatic

Hydrocarbons (2) 14.3: Aliphatic

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*Hydrocarbons (10) 14.4: Derivatives
of Hydrocarbons (4) 14.5: Synthetic
Polymers (8) 14.6: Biochemistry ;
14: Multiple Choice (12) 14: Short
Answer (19) 14: Applying Your*

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Knowledge ; 14: Extra Problems (1)
Chapter 15: Place and Time

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**Answer Key Chapter 14 -
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*14.1 History of Chemotherapy and
Antimicrobial Discovery; 14.2
Fundamentals of Antimicrobial*

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*Chemotherapy; 14.3 Mechanisms of
Antibacterial Drugs; 14.4
Mechanisms of Other Antimicrobial
Drugs; 14.5 Drug Resistance; 14.6
Testing the Effectiveness of*

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*Antimicrobials; 14.7 Current
Strategies for Antimicrobial*

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Discovery; Summary

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Heat | Physics

Heat is energy transferred solely due to a temperature difference. Any energy unit can be used for heat transfer, and the most common are

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*kilocalorie (kcal) and joule (J).
Kilocalorie is defined to be the
energy needed to change the
temperature of 1.00 kg of water*

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between 14.5°C and 15.5°C.

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Galaxy

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approximate temperature of the surface of the Sun? The temperature of the photosphere is between 4000 and 6000 degrees Celsius. 1. b) The

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corona is...

14.2 Temperature Change and

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Heat Capacity – College ...

Because heat transfer between the pan and water takes place rapidly, the mass of evaporated water is negligible and the magnitude of the

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heat lost by the pan is equal to the heat gained by the water. The exchange of heat stops once a thermal equilibrium between the pan

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and the water is achieved.

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Chemistry, 7e (Wade) Chapter 14
Ethers, Epoxides, and Sulfides 1)*

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bond angle in dimethyl ether?

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Chapter 14 Intermolecular Forces

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Dipole Example 2.

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*Experiment 14: Heat Effects and
Calorimetry Purpose: To fully
understand the effects of heat and to
calculate the specific heat of an*

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*unknown. Experimental Procedure:
Part A: Specific Heat 1. From your
instructor obtain a calorimeter, a
sensitive thermometer, a sample of
metal in a large stoppered test tube.*

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*The calorimeter consists of two
nested expanded polystyrene coffee*

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cups fitted with a ...

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Answers

*Chapter 14 and 15 Study Guide
Answers. Section 14-1.*

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Biogenesis is the principle that all

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from nonliving things.

BAM Chapter 14: Bacillus cereus |

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FDA

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BAM for preparation of food
homogenate Pipets, 1, 5, and 10 ml,
graduated in 0.1 ml units Glass*

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spreading rods (e.g., hockey stick) ...

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2/9/2019 · Answer. When we breathe out, we exhale carbon dioxide gas along with water vapours. The water vapour on colliding with the surface

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of the glass gets ...

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14 HEAT AND HEAT

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TRANSFER METHODS

Energy can exist in many forms and heat is one of the most intriguing.

Heat is often hidden, as it only exists when in transit, and is transferred by

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a number of distinctly different methods. Heat transfer touches every aspect of our lives and helps us understand how the universe functions. It **CHAPTER 14 | HEAT**

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AND HEAT TRANSFER METHODS
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CHAPTER 14 THE CLAUSIUS- CLAPEYRON EQUATION

*$H_1 + TS_1 = H_2 + TS_2$ 14.1 or $T(S_2 + S_1) = H_2 - H_1$, 14.2 in which the
enthalpy and entropy are specific.*

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The left hand side is the specific latent heat of vaporization, and we already knew from Chapter 9 that this was equal to the difference in the specific enthalpies of liquid and

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vapour.

Chapter 14 | Helen's Guide To The

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approximate temperature of the surface of the Sun? The temperature of the photosphere is between 4000 and 6000 degrees Celsius. 1. b) The corona is at a very high temperature,

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but produces little 'heat'.

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Chapter 14 Intermolecular Forces

Answers: A) nonpolar, LDF; B) polar, DDF; C) polar; HBF; D) nonpolar, LDF; E) polar, IDF Image from "Salts", by OpenStax Anatomy

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Answer Keys - HONORS

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CHEMISTRY

*SG 14.3 Ch.14 Review Chapter 16
Measuring Heat Flow One vs. Two
System Problems SG 16.1 & 16.2
Calorimetry Lab Thermochemical*

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*Equations Hess's Law Worksheet SG
16.4 SG 16.3 & 16.5 Gibbs Free
Energy Chapter 16 Review
Reviewing Vocabulary Chapter 19
Section 19.1 Review ? SG 19.1 &*

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19.2 ? ? Understanding Logarithms
? SG 19.3 Determining pH ...

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CHAPTER 14**

*1. The specified equipment or
appliance utilizes multistage
technology or variable refrigerant*

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*flow technology and the loads
calculated in accordance with the
approved heating and cooling
calculation methodology are within
the range of the manufacturer's*

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*published capacities for that
equipment or ...*

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CHAPTER 14

About this chapter: Chapter 14 addresses requirements for exterior walls of buildings. Minimum standards for wall covering

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*materials, such as material
performance and fire resistance,
installation of wall coverings and the
ability of the wall to provide weather*

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protection are provided.

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Chapter 16 HEAT EXCHANGERS

the same flow area. In compact heat exchangers, the two fluids usually move perpendicular to each other.

16-3C A heat exchanger is classified

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*as being compact if $\alpha > 700 \text{ m}^2/\text{m}^3$
or $(200 \text{ ft}^2/\text{ft}^3)$ where α is the ratio of
the heat transfer surface area to its*

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volume which is ...

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