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Chapter 5: Periodic Classification ...

The main credit for classifying elements goes to Dmitri Ivanovich  
Mendelée'v, a Russian chemist. He was the most important  
contributor to the early development of a Periodic Table of

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elements wherein the elements were arranged on the basis of their fundamental property, the atomic mass, and also on the similarity of chemical properties.

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www.ncerthelp.com (Visit for all ncert solutions in text and videos, CBSE syllabus, note and many more) Science Class 10 Notes for Periodic Classification of Elements 1. Classification means identifying similar species and grouping them together. 2. Lavoisier divided elements into two main types known as metals and non-metals. 3.

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www.embibe.com . Periodic Classification of Elements . Back of Chapter Questions . 1. Which of the following statements is not a correct statement about the trends when going from left to right across the periods of periodic Table. (A) The elements become less metallic in nature.

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**CLASSIFICATION OF ELEMENTS AND PERIODICITY IN PROPERTIES** 71 3.2 GENESIS OF PERIODIC CLASSIFICATION Classification of elements into groups and

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development of Periodic Law and Periodic Table are the consequences of systematising the knowledge gained by a number of scientists through their observations and experiments. The German chemist, Johann ...

NCERT Solution for class 10 Science Chapter 5 Periodic Classification of Elements Si-Si<sub>2</sub>O<sub>4</sub> or SiO<sub>2</sub> Ba<sub>2</sub>O<sub>2</sub> Oxygen is a member of group VI A in Mendeleev's periodic table. Its valency is 2. Similarly, the valencies of all the elements listed can be predicted from their respective groups. This will help in writing the formulae of their oxides.

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NCERT Exemplar for Class 10 Science Chapter 5 Periodic Classification of elements Explanation: Element with atomic no. 14 is Silicon and is a metalloid. It forms acidic oxide , thus behaving as a non- metal and also forms covalent halide, thus acting as a metal.

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23.

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