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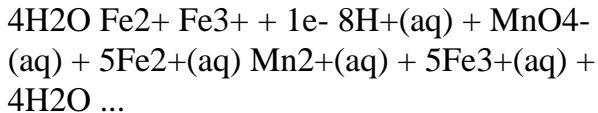
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Hydrogen Peroxide Purpose: To analyze the**

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concentration of hydrogen peroxide by using potassium permanganate in titration to be able to check the quality Pre-Lab

Questions: 1.  $5e^- + MnO_4^- + 8H^+ \rightarrow Mn^{2+} +$

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1.00 mL of hydrogen peroxide and put it in

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the Erlenmeyer flask. Measure 10mL of sulfuric acid and add to the flask and swirl. Measure 20 mL of distilled water and add to the flask.... Analysis of Hydrogen



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17/2/2010 · Experiment average:  $(3.253 + 2.551 + 3.423) \div 3 = 3.075$  %. The actual percent of hydrogen peroxide is 3%, so the difference between the experimental result

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and theoretical result is 0.075%. 6. If an insufficient amount of acid is added in step 9, some of the  $\text{MnO}_4^-$  ...

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