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(BSc, Ancillary
Chemistry, SemIV, 2014-
2015) 1. Is it possible to
recover copper metal from
the solution of copper
sulphate using Iron scrap?

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Discuss, Given: (4 marks)
 $E^0(\text{Fe}^{2+}/\text{Fe}) = -0.41$
volt ; $E^0(\text{Cu}^{2+}/\text{Cu}) =$
0.34 volt 2. Write short
note on rusting of iron and
its prevention (4 marks)
(BSc, Ancillary
Chemistry, SemIV, 2016-
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solution is available to use
as the titrant. The student
uses a 25.00 mL
volumetric pipet to deliver
the propanoic acid solution
to a clean, dry flask. After

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adding an appropriate indicator to the flask, the student titrates the solution with the 0.173 M NaOH, reaching the end point after 20.52 mL of the base solution ...

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pH of a solution prepared

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by mixing the . 50.0 mL sample of . 0.20 M $\text{CH}_3\text{CH}_2\text{COOH}$ with a . 50.0 mL 0.20 M NaOH is . 7.00. (ii) If the . pH of a hydrochloric acid solution is the same as the . pH of a propanoic acid solution, then the molar . sample of

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. concentration of the hydrochloric acid solution must be less than the molar concentration ...

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examination was the third exam for the current VCE Chemistry Study Design and covered the designated key knowledge and key skills from Units 3 and 4. Demonstrating skills and understanding of chemical ideas in examinations is

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solution ...

pH of a solution prepared
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sample of . 0.20 M CH₃
CH₂ COOH with a . 50.0

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mL 0.20 M NaOH is .
7.00. (ii) If the . pH of a
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