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Chemistry Semester 2

Midterm Review Answers

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Chemical Reactions = 5  
Questions (10%) Be able  
to identify the type of  
chemical reaction given a  
chemical reaction. Be able  
to identify the chemical  
reaction given its type. Be  
able to balance a chemical  
reaction. Understand the



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requirements for chemical  
reactions to occur.

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Ionic and Covalent  
Compounds : a. Ionic  
Covalent nonmetal + ...

**Chemistry Midterm  
Study Guide 4.1.1**

Describe Democritus's  
ideas about atoms. 4.1.2  
Explain Dalton's atomic

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theory. 4.1.3 Identify what instrument is used to observe individual atoms.

4.2 Structure of the Nuclear Atom, pp.

104–109 4.2.1 Identify three types of subatomic particles.

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Chemistry Semester 2

Midterm Exam Study

Guide Name: \_\_\_\_\_ Unit

5: Stoichiometry and  
moles Chapters 10, 11, 14

Teacher initial: \_\_\_\_\_

Moles, Molar mass,

Empirical Formula,

Molecular Formula. 1.

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What is the mass of  $8.4 \times 10^{-3}$  mol of zinc (II) chloride? 2. How many moles of calcium nitrate are in 11.7 g? 3.

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**GUIDE PAGE 1 OF 8**

**Freehold Regional High  
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**PRINCIPLES OF LAB  
CHEMISTRY MIDTERM  
EXAMINATION STUDY  
GUIDE About the Exam**

**The Lab Chemistry  
midterm exam consists of**

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45 multiple choice questions. With each question, five choices will be provided. You must choose the best answer.

20/2/2019 · CHEM 120  
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Designing an Experiment •

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What makes a good hypothesis? • Design experiments by changing only a single variable – why is this important? • What are the differences between an independent, dependent and controlled variable? Units and



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density • Know metric units for mass, length, volume, time and temperature. • Recognize prefixes and know what they mean ...

Midterm study guide Read each question. Follow the

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instructions to answer the questions. 1. Jenna is studying how matter changes during reactions. She wants to compare the weight of her materials before and after the reaction. Which tool should she use? Circle the

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letter of the correct answer.

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Midterm Review 2016.  
125 Terms.

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Chemistry Chapter 10 -

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NMR. 21 Terms.  
Rebekah\_Belasco.

This guide will offer an overview of the main tested subjects, along with sample AP multiple-choice questions that look like the questions you will

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see on test day. Atomic  
Structure and Properties  
Around 7-9% of questions  
on your AP Chemistry  
exam will cover Atomic  
Structure and Properties.

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Describe Democritus's ideas about atoms. 4.1.2  
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4.2.1 Identify

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three types of subatomic  
particles.

**Chemistry Midterm  
Study Guide** Chemical  
Reactions = 5 Questions  
(10%) Be able to identify  
the type of chemical  
reaction given a chemical

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reaction. Be able to identify the chemical reaction given its type. Be able to balance a chemical reaction. Understand the requirements for chemical reactions to occur.

**CHEM 344 Midterm**



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Exam Study Guide

Purification techniques

Acid-Base Extraction ...

Nucleophilic substitution  
reactions (also covered in  
CHEM 343) identification  
of S N 1 and S N 2  
reactions & appropriate  
substrates and conditions

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for each; mechanisms of S  
N 1 and S N 2 reactions.

1 Chemistry Semester 2  
Midterm Exam Study  
Guide Name: \_\_\_\_\_ Unit  
5: Stoichiometry and  
moles Chapters 10, 11, 14  
Teacher initial: \_\_\_\_\_

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Moles, Molar mass,  
Empirical Formula,  
Molecular Formula.

Midterm study guide Read  
each question. Follow the  
instructions to answer the  
questions. 1. Jenna is  
studying how matter

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changes during reactions.  
She wants to compare the  
weight of her materials  
before and after the  
reaction. Which tool  
should she use? Circle the  
letter of the correct answer.

Regents Chemistry Part A

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Midterm Test 1 - 50

Multiple Choice 17. An element that is malleable and a good conductor of heat and electricity could have an atomic number of  
1) 16 3) 29 2) 18 4) 35 18.

The transfer of electrons from sodium atoms to

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chlorine atoms results in  
the formation of 1)  
metallic bonds 3) nonpolar  
bonds

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GUIDE FOR 8TH  
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Welcome to chemistry.

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Chemistry is the science of matter and its interactions. Everything around you is made of atoms – atoms and their chemical combinations, molecules. Everything you can see, touch, smell, and taste is made up of chemicals. So,

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the science of chemistry is a very wide ranging science.

This guide will offer an overview of the main tested subjects, along with sample AP multiple-choice questions that look



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like the questions you will see on test day. Atomic Structure and Properties Around 7-9% of questions on your AP Chemistry exam will cover Atomic Structure and Properties.

List of AP Chemistry Tips

Read Chemistry  
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ebooks  
as well as a How-to Guide  
for studying. It features  
information from the  
Albert Blog. If youre  
looking for additional  
resources, be sure to  
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Chemistry, 2015 The  
midterm exam will include  
content from Chapters  
1,2,3,4,5,6 of your  
textbook. When studying,

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you should review how the terms below relate to your tests and quizzes from each of these chapters.

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Exam Study Guide  
Purification techniques  
Acid-Base Extraction ...

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Nucleophilic substitution reactions (also covered in CHEM 343) identification of S<sub>N</sub>1 and S<sub>N</sub>2 reactions & appropriate substrates and conditions for each; mechanisms of S<sub>N</sub>1 and S<sub>N</sub>2 reactions.



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aluminum to xenon, we  
explain the properties and  
composition of the  
substances that make up  
all ...

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Welcome to chemistry.  
Chemistry is the science of  
matter and its interactions.  
Everything around you is  
made of atoms – atoms  
and their chemical  
combinations, molecules.

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Everything you can see, touch, smell, and taste is made up of chemicals. So, the science of chemistry is a very wide ranging science.

This guide will offer an overview of the main

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tested subjects, along with sample AP multiple-choice questions that look like the questions you will see on test day. Atomic Structure and Properties  
Around 7-9% of questions on your AP Chemistry exam will cover Atomic

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Structure and Properties.

List of AP Chemistry Tips  
as well as a How-to Guide  
for studying. It features  
information from the  
Albert Blog. If youre  
looking for additional  
resources, be sure to

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Chemistry 221 Chapter  
Guide: Midterm Exam II

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Chemistry 221 at Mt.  
Midterm Exam II is  
approaching, and this is  
the time to study hard to  
prepare for it. Here are  
some hints to get you

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Chapter 3, Chapter 4 &  
Chapter 5.

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semester chapters, be sure  
to also review the  
additional practice

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questions from each  
chapter that were attached  
to Semester 1 Study

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REVIEW: REFER TO

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in the book and check your

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answers in the back

(Appendix E, page R82) o  
p. 30 #s 26 & 27

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