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Chemistry-Nail Lab By: Tera Nguyen Observations: Day 1-the water turned more of a darker color-the nail got thicker Day 2-the water evaporated-very thick Data: Mass 250 mL beaker 106.974 g Mass 250 mL beaker + copper (II) chloride 112.264 g Mass nails before reaction 5.29 g Mass nails after reaction 4.50 g Mass 250 mL beaker + dry copper 107.84 g Calculations: 1. ...

Chemistry - Nail Lab Purpose The purpose is to determine the ratio of copper produced to iron consumed in a replacement reaction. Procedure Day 1 1. Label, then mass a 250 mL beaker. 2. Put 50.0 mL of copper (II) chloride in the beaker. 3. Mass 2 or 3 nails together to ± 0.01 g. 5. Place the nails in the copper chloride solution.

9/9/1985 · Name: _____ Date: _____ Unit 7 Chemistry - Nail Lab Purpose After you have completed the lab, think about what was the overall point for performing the lab. Procedure Day 1 1. Mass a clean & labeled _____ mL beaker. 2. Put between 6.0 - 8.5 g of copper (II) chloride in the beaker.

Los Angeles City College Chemistry 51 Fall 2005 3093 Experiment 3 Introduction to Density INTRODUCTION The purpose of this experiment is to understand the meaning and significance of the density of a substance. Density is a basic physical property of a homogeneous substance; it is an intensive property, which

It can affect the habitat in the water. The fishes may die easily because of our bad manners. For this experiment, the solution produce two types of smell and one is odorless. Organic Synthesis : Formation Of An Ester Lab Report Page 10 8.0 REFERENCE 1) 2006, ORGANIC CHEMISTRY, McGraw Hill ...

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Recognition of equipment found in your workstation and lab along with their use. II. Calculations A. The basic calculations used in this lab are important for future work in chemistry and other sciences. Mastery of the following calculations and methods is expected: 1. (Percent by mass Experiments 2, 6). 2. (Density Experiments 1, 2). 3.

A lab apron or coat will prove a good investment for protecting clothing. Learn the location of the safety equipment: fire extinguisher, safety shower, and eye wash fountain. Injuries/accidents should be immediately reported to your instructor. 2. Lab work is to be done on an individual basis unless you are instructed otherwise.

You will find answers and maths skills required for the practicals in the back of the book. INTRODUCTION M01-iASAL_CHEM_LAB_44716_001-076.indd 2 08/06/2018 11:44 Uncorrected proof, all content subject to change at publisher discretion. Not for resale, circulation or distribution in ...

Lab 5: Battery Lab Report Due May 18, 2011, in class 2 Procedure I. The Lemon Battery A lemon battery is made using a lemon and two metallic electrodes. The juice in the lemon is the electrolyte in the battery. When the electrodes are connected by a wire, a chemical reaction occurs at each electrode.

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