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a consistent mental model of matter as discrete particles.

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Problems On each of the problems below,
start with the given P, V, T, or n; then make
a decision as to how a change in P, V, T, or
n will affect the starting quantity, and then
multiply by the appropriate factor.

Unit 2 Worksheet 3 – PVTn Problems On
each of the problems below, start with the
given P, V, T, or n; then make a decision as
to how a change in P, V, T, or n will affect
the starting quantity, and then multiply by
the appropriate factor. Draw particle
diagrams of the initial and final conditions.

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1. A sample of gas occupies 150 mL at 25 °C.

Unit 2 Worksheet 5 – PVTn Gas

Relationships Practice On your own paper, make a chart to record the initial values, final values, and effects for P, V, T, or n. Set up a problem, multiplying by the appropriate factor(s). Record the answer with the correct number of significant figures and units and box the answer. 1.

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multiply by the appropriate factor.

Unit 2 Worksheet 3 – PVTn Problems On each of the problems below, start with the given P, V, T, or n; then make a decision as to how a change in P, V, T, or n will affect the starting quantity, and then multiply by the appropriate factor. Draw particle diagrams of the initial and final conditions.

1. A sample of gas occupies 150 mL at 25 °C.

Unit 2 Worksheet 5 – PVTn Gas

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with the correct number of significant figures and units and box the answer. 1.

Name: Abdirahman Mohamed Date

10/28/2020 Unit 2 Worksheet 3 – PVTn

Problems On each of the problems below, start with the given P, V, T, or n; then make a decision as to how a change in P, V, T, or n will affect the starting quantity, and then multiply by the appropriate factor.

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start with the given P, V, T, or n...

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Modeling Chemistry 1 U2 ws 3 v2.1 Name _____ Date Pd _____ Unit 2 Worksheet 3 – PVTn Problems On each of the problems below, start with the given P, V, T, or n; then make a decision as to how a change in P, V, T, or n will affect the starting quantity, and then

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Unit 2 Worksheet 3 – PVTn Problems On
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A closed flask of air (0.250 L) contains 5.0 "puffs" of particles. The pressure probe on the flask reads 93 kPa. A student uses a syringe to add an additional 3.0 "puffs" of air through the stopper. Find the new pressure inside the flask. A 350 mL sample of gas has a temperature of 30 °C and a

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pressure of 1.20 atm.

2]2[O 3] 0 c. Determine the value and units of the rate constant, k . plug and chug using the rate law & data from exp't 1 and solving for k , we get $k = 2.36 \text{ mol.L}^{-1} \cdot \text{s}^{-1}$.

Consider the following mechanism. $A_2 + B_2 \rightarrow R + C$ (slow) $A_2 + R \rightarrow C$ (fast) a. Write the overall balanced chemical equation. $2 A_2 + B_2 \rightarrow 2 C$ b.

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