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Ionic and metallic bonds P.2/11 Formation of ionic bonds From atoms to ions Atoms can obtain the stable electronic arrangements of atoms of noble gases by gaining or losing electrons. A simple ion forms when an atom either loses or gains one or more electrons. It is either positively or negatively charged. [An ion can also be formed by two or more atoms joined together by covalent bonds first.

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Introduction to Ionic Bonds The forces that hold matter together are called chemical bonds. There are four major types of bonds. We need to learn in detail about these bonds and how they influence the properties of matter. The four major types of bonds are: I. Ionic Bonds III. Metallic Bonds II. Covalent Bonds ...

Chemistry chapter 7 ionic and metallic bonding worksheet answers CHAPTER 6 Chemical Bonding SECTION 1 Introduction to Chemical Bonding OBJECTIVES 1. Define Chemical bond. 2. Explain why most atoms form chemical bonds. 3. Describe ionic and covalent bonding.. 4.

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