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# Reading Fundamental Concepts Inorganic Chemistry Antivi Combinations 1.

## Introduction 1.1.

Development of atomic theory Atomic theory is the study of atoms and the force which hold them together. Philosophers and scientists did not arrive at the atomic theory alone.

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geometry and structure,  
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Group Elements and



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Chemistry of Transition  
Elements. The point  
however is that this e-book  
does present the facts in a  
very systematic way. We  
have a vividly baconian  
philosophy about  
Chemistry, but particularly  
in General chemistry

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concepts. We are  
convinced that inorganic  
chemistry sans facts as  
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present this book,

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serve the needs of the  
students. The book has  
been written keeping in  
mind the general weakness  
in understanding the  
fundamental concepts of  
the topics. The book is self-  
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“Teach Yourself” style. It  
is based on question-  
answer pattern.

Coordination compound  
(complex)- basics •  
Central atom is bound to  
unexpectedly large  
number of ligands •

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Usually discrete species in  
solution and solid •

Examples:  $K_4[Fe(CN)_6]$ ,  
 $CoCl_2 \cdot 6 H_2O$  •

Properties of central atoms  
(transition metals): • Large  
charge/radius ratio •

Variable oxidation states  
(d-electrons available) •

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oxidation states, s-  
electrons are ...

A text book of inorganic  
chemistry. This book is  
divided in to three parts.  
Part I: contains a brief  
sketch of the fundamental

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principles and theories  
upon which the science of  
modern chemistry is built.  
Part II: consists of the  
study of the four typical  
elements, hydrogen,  
oxygen, nitrogen, and  
carbon, and of their more  
important compounds.



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chemical industry and  
environmental chemistry.

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Here is a detailed list of our topics (the chapter numbers are from our textbook Shriver & Atkins' Inorganic Chemistry 5th edition):  
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1. Inorganic chemistry – a general

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CONCEPTS OF  
CHEMISTRY** Chemistry  
is the science of molecules  
and their transformations.

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It is the science not so  
much of the one hundred  
elements but of the infinite  
variety of molecules that  
may be built from them.  
Roald Hoffmann Science  
can be viewed as a  
continuing human effort to  
systematise knowledge for

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