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Answer the following questions about Practice Problem 2. Sulfur burns in oxygen to form sulfur dioxide. Write a skeleton equation for this chemical reaction. Analyze Step 1. Write the formula for each reactant and each product. Include the common STP state of each substance. Reactants Products Solve Step 2.

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Education U2-33 UIT 2 REASOIG WITH
LIEAR EUATIOS AD IEUALITIES 1
Instruction ... Read the problem statement
carefully. 2. Reread the scenario and make a
list of the known quantities. Alexis has
saved \$400. She makes \$12 an hour. She

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needs at least \$1,100.

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PROBLEM 3 (page 327) 3. Balance each equation. a. $\text{AgNO}_3 + \text{H}_2\text{S} \rightarrow \text{Ag}_2\text{S} + \text{HNO}_3$

b. $\text{Zn}(\text{OH})_2 + \text{H}_3\text{PO}_4 \rightarrow \text{Zn}_3(\text{PO}_4)_2 + \text{H}_2\text{O}$

a. $\text{AgNO}_3 + \text{H}_2\text{S} \rightarrow \text{Ag}_2\text{S} + \text{HNO}_3$

Analyze Step 1. Identify the relevant concepts. Count the number of atoms of each element on both sides of the skeleton

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equation.

10/12/2017 · Sample Problem 1: Decide whether each sequence is arithmetic. The explicit formula for the general term of an arithmetic sequence is $a_n = a_1 + (n - 1)d$, where a_1 is the first term, n is the number of terms, and d is the common difference. - the general term or n th term

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Solve Step 2.

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Balancing Equations: Practice Problems 1.

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Balance each of the following equations. (a) $\text{Fe} + \text{Cl}_2 \rightarrow \text{FeCl}_3$ (b) $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$
(c) $\text{FeBr}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + \text{HBr}$
(d) $\text{C}_4\text{H}_6\text{O}_3 + \text{H}_2\text{O} \rightarrow \text{C}_2\text{H}_4\text{O}_2$ (e) $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ (f) $\text{C}_4\text{H}_{10}\text{O} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ (g) $\text{C}_7\text{H}_{16} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ (h) $\text{H}_2\text{SiCl}_2 + \text{H}_2\text{O} \rightarrow \text{H}_8\text{Si}_4\text{O}_4 + \text{HCl}$ (i) $\text{HSiCl}_3 + \text{H}_2\text{O} \rightarrow \dots$

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PROBLEM 3 (page 327) 3. Balance each equation. a. $\text{AgNO}_3 + \text{H}_2\text{S} \rightarrow \text{Ag}_2\text{S} + \text{HNO}_3$
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Analyze Step 1. Identify the relevant concepts. Count the number of atoms of each element on both sides of the skeleton

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equation.

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whether each sequence is arithmetic. The
explicit formula for the general term of an
arithmetic sequence is $a_n = a_1 + (n - 1)d$. a_1 the
first term n the number of term d the
common difference a_n - the general term or
 a_n term

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PROBLEM 3 (page 327) 3. Balance each equation.
a. $\text{AgNO}_3 + \text{H}_2\text{S} \rightarrow \text{Ag}_2\text{S} + \text{HNO}_3$
b. $\text{Zn}(\text{OH})_2 + \text{H}_3\text{PO}_4 \rightarrow \text{Zn}_3(\text{PO}_4)_2 + \text{H}_2\text{O}$

Analyze Step 1. Identify the relevant concepts. Count the number of atoms of each element on both sides of the skeleton equation.

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$a = 2$ $b = 4$ Since the problem is given in
months, to find x , you need to convert 2
years into 6-month time periods ... are in 1
year. There are 2. Multiply that by 2 for
each year. Therefore, there are four 6-
month time periods in 2 years. (continued)

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Guided Practice 4.1.1 SWB p. 9. GSE
Algebra I Teacher Resource 4.1.1 Walch
Education U4-11 UIT ...

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Balance each of the following equations. (a)
 $\text{Fe} + \text{Cl}_2 \rightarrow \text{FeCl}_3$ (b) $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$
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(d) $\text{C}_4\text{H}_6\text{O}_3 + \text{H}_2\text{O} \rightarrow \text{C}_2\text{H}_4\text{O}_2$ (e) C_2H_4

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+ O₂ ??? CO₂ + H₂O (f) C₄H₁₀O + O₂ ???
CO₂ + H₂O (g) C₇H₁₆ + O₂ ??? CO₂ +
H₂O (h) H₂SiCl₂ + H₂O ??? H₈Si₄O₄ +
HCl (i) HSiCl₃ + H₂O ?? ...

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medicine on 4,250 people. It is effective for
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