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pair of electrons and has achieved a stable noble gas configuration. 3.

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Hydrogen atoms always have one covalent bond and no lone pairs, and oxygen atoms usually have two covalent bonds and two lone pairs. d. ethylene (or ethene), C_2H_4 (used to make polyethylene) Carbon atoms form four bonds with no lone pairs, and hydrogen atoms form one bond with no lone pairs. To achieve these bonding patterns, there must be a

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SECTION 1 continued 6. Some binary compounds are ionic, others are covalent. The type of bond favored partially depends on the position of the elements in the periodic table.

Name Class Date I Concept Review continued 9. The empirical formula of a substance is BH_3 . The experimental molar mass is 27.67 g/mol. Determine the molecular formula of the compound. 10. The empirical formula of a compound is CH . The experimental molar mass is 26.04 g/mol. Determine th...

the modern concepts regarding ion-formation by electron transfer and the formation of ionic crystalline compounds. His views have proved to be of great value in the understanding and systematisation of the ionic compounds. At the same time he did recognise the fact that a large number of compounds did not fit into these concepts. 4.1.1 ...

chemistry and basic concepts for learning inorganic chemistry are presented in the first three chapters. Inorganic chemistry is of fundamental

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