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1. How many atoms are in

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3.5 grams of silicon? How

many formula units are in

32.6 grams of potassium

oxide? How many

molecules are in 0.25

grams of

dinitrogenpentoxide?

Episode 701? atoms =

3.5g Si 1mol Si $6.022 \times$

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10. 23 . atoms. 28.1g Si 1

mol Si = $7.5 \times 10. 23.$

atoms Si ...

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Stoichiometry e study of
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+ 02 2 Mgo Name

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equations cc L, ch
cpnöducc- The Ex.
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involved in the When
elemental aluminum reacts
with elemental iodine,
aluminum iodide is
produced.

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3.5 grams of silicon? How
many formula units are in
32.6 grams of potassium

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oxide? How many
molecules are in 0.25
grams of

dinitrogenpentoxide?

Episode 701? atoms =

$3.5\text{g Si} \frac{1\text{mol Si}}{28.1\text{g Si}} \times 6.022 \times$

$10^{23} \text{ atoms} = 7.5 \times 10^{23}$

$\text{mol Si} = 7.5 \times 10^{23}$

atoms Si ...

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+ 02 2 Mgo Name
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cpnöducc- The Ex.
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involved in the When
elemental aluminum reacts
with elemental iodine,
aluminum iodide is
produced.

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called , Gas molecules are
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Acharya Bohr's Energy

Levels • Electrons in

certain energy levels •

Lower energy levels:

closer to nucleus • Higher

energy levels: further from

nucleus • Ground State:

electron in lowest energy

level possible Excited

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Atom • Atom has absorbed energy • Excited state is unstable • Atom soon emits same amount of energy absorbed • energy ...

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that each element has a
unique atomic number.

.arranged elements by
atomic mass number. .now
all elements fit into place
based on properties. Term.

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