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periodic table. Name two of the sets of elements, more for bonus points. Alkali/Alkaline Earth Metals 1. Color the alkali metals orange, and the alkaline earth metals green on the periodic at the back of the packet. Draw a key at the bottom of the page that identifies the meaning of each color. Continue expanding this key

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as you add to the table.

Periodic Table Study Guide Directions: This study guide is an excellent review for the test. If you have questions regarding the study guide or test please see the teacher before the class period of the test. **Key Terms Directions:** Define the following terms. 1. atomic number: 2. atomic mass: 3. symbol: 4. luster: 5. ductile: 6. malleable: 7 ...

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atomic number 17 is? 5. List the symbols for two transition metals.
6.

Periodic Trends Matching Match each item with the correct statement below. a. electronegativity f. periodic law b. ionization energy g. atomic mass c. atomic radius h. period d. metal i. group e. transition metal j. electrons ____ 1. horizontal row in the periodic table ____ 2. vertical column in the periodic table

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This is a SUPER easy guide on Periodic table of elements. You will come to know everything about the Periodic table, including; History of Periodic Table, Periodic Table with Names, symbol, atomic mass, atomic number, electron configuration and lots more...

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atomic number: 2. atomic mass: 3. symbol: 4. luster: 5. ductile: 6.
malleable: 7 ...

The Periodic Table of the Elements 1 H Hydrogen 1.00794 2 He
Helium 4.003 3 Li Lithium 6.941 4 Be Beryllium 9.012182 5 B
Boron 10.811 6 C Carbon 12.0107 7 N Nitrogen 14.00674 8 O
Oxygen 15.9994 9 F Fluorine 18.9984032 10 Ne Neon 20.1797 11
Na Sodium 22.989770 12 Mg Magnesium 24.3050 13 Al
Aluminum 26.981538 14 Si Silicon 28.0855 15 P Phosphorus ...

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energy g. atomic mass c. atomic radius h. period d. metal i. group e.
transition metal j. electrons ____ 1. horizontal row in the periodic
table ____ 2. vertical column in the periodic table

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3 Li Lithium 6.941 4 Be Beryllium 9.012182 5 B Boron 10.811
6 C Carbon 12.0107 7 N Nitrogen 14.00674 8 O Oxygen 15.9994
9 F Fluorine 18.9984032 10 Ne Neon 20.1797 11

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Na Sodium 22.989770 12 Mg Magnesium 24.3050 13 Al
Aluminum 26.981538 14 Si Silicon 28.0855 15 P Phosphorus ...

The Periodic Table of Elements 7 Li lithium 3 23 Na sodium 11 39
K potassium 19 85 Rb rubidium 37 133 Cs caesium 55 [223] Fr 87
* The Lanthanides (atomic numbers 58 – 71) and the Actinides
(atomic numbers 90 – 103) have been omitted. Relative atomic
masses for Cu and Cl have not been rounded to the nearest whole
number. . francium 12 Be Key 9 beryllium 4 relative atomic mass

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atomic symbol ...

Study Guide for Atoms & Periodic Table Unit Test Study all warm ups Know how to use an elements name, symbol, atomic number, protons, electrons or atomic mass to find it on the periodic table. Know how to find an elements name, symbol, atomic number, protons, electrons, neutrons and atomic mass if you are given one of them.

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Atomic Structure and the Periodic Table Study Guide Below are the expectations for each point level for the three Atomic Structure and the Periodic Table objectives. Objective 1: Students will know the basic structure of an atom. 1 point – (Basic)

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atomic masses of the elements and their physical and chemical properties. Among chemical properties,

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