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Solubility Curves Worksheet Answers

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Solubility Curves Study the solubility
curves in the figure, and then answer the
questions that follow. 1. What relationship
exists between solubility and temperature
for most of the substances shown? 2. a.
What is the exception? NN t i b. What
general principle accounts for this ex-
ception? 6 6" 3. a. Approximately how
many grams of NaNO_3

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harmful downloads. Unit 12 review worksheet part 1 solubility curves use your solubility curve graph to answer 14. At 55 °C a saturated solution would contain 120 g of solute NaNO_3 .

For questions 11 – 12, show your work and circle your final answer. 11. If 115 g KNO_3 are added to 100 g of water at 35 °C, how many grams do not dissolve? 12. What mass of KCl would be needed to form a saturated solution if the KCl was dissolved in 200 g of water at 80 °C?

12. Assume that a solubility curve for a gas such as carbon dioxide, at one atmosphere of pressure, was plotted on the solubility curve graph. Reading from left to right, this

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curve would a. slope upward b. slope downward c. go straight across

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31/1/2020 · Solubility curve worksheet key use your solubility curve graphs provided to answer the following questions. At 55 c a saturated solution would contain 120 g of solute NaNO_3 . What mass of KCl would be needed to form a saturated solution if the KCl was dissolved in 200 g of water at 80 c. Or to determine the least or most soluble substances 6.

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temperatures on the graph only go from 0°
C to 100° C ? Water changes to a gas at
 100° C and will no longer dissolve solute as
a gas. 2. Which substance is most soluble at
 60° C? potassium iodide 3. Which two
substances have the same solubility at 80° C

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?_ sodium chloride, potassium chlorate

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Solubility Curve Worksheet You'll notice that for most substances, solubility increases as temperature increases. As discussed earlier in solutions involving liquids and solids typically more solute can be dissolved at higher temperatures. Can you find any exceptions on the graph?_____ Here's an example of how to read the graph.

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so they will come in handy.

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Solubility graph with key. Steep curves reflect the greatest effect. Which is the least soluble at 40 °c. Supersaturated solutions and solubility curves will be a part of the quiz. Use the graph to answer the following two questions. Degrass celsius and grams of ...

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Using a solubility curve, determine the amount of each solute that can dissolve in 100 g of water at the given temperature. 140 t 30 120 ... Use the graph to answer the questions below. If the question requires a calculation, show work. q 150 140 KI 130 120 110 100 90 80 70 Na₂SO₄ 50 30 20 NaNO₃ NH₃ gases

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Solubility Graph Worksheet 1. Why do the temperatures on the graph only go from 0°C to 100°C ? Water changes to a gas at 100°C and will no longer dissolve solute as a gas. 2. Which substance is most soluble at 60°C ? potassium iodide 3. Which two substances have the same solubility at 80°C ? sodium chloride, potassium chlorate

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that for most substances, solubility
increases as temperature increases. As
discussed earlier in solutions involving
liquids and solids typically more solute can
be dissolved at higher temperatures. Can
you find any exceptions on the
graph?_____ Here's an example of how to

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read the graph.

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