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answers the question. _____

1. The first step in most stoichiometry problems is to _____. a. add the coefficients of the reagents c. convert given quantities to volumes b. convert given quantities to moles d. convert given quantities

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36. a. Two formula units
 KClO_3 decom- pose to
form two formula units

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O₂. b. Four molecules
NH₃ ...

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Section 11.1 Defining

Stoichiometry pages

368–372 Practice

Problems pages 371–372

1. Interpret the following
balanced chemical equa-
tions in terms of particles,

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moles, and mass. Show
that the law of
conservation of mass is

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Problems pages 371–372

1. Interpret the following balanced chemical equations in terms of particles, moles, and mass. Show that the law of conservation of mass is

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