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question. 1) A chemical reaction that absorbs heat from the surroundings is said to be _____ and has a _____ ΔH at constant pressure. A) endothermic ...

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1. Consider 2 metals, A and B, each having a mass of 100 g and an initial temperature of 20 °C. The specific heat of A is larger than that of B. Under the same heating conditions, which metal would take longer to reach 21 °C? Explain your reasoning. 2.

Thermochemistry Example Problems

Recognizing Endothermic & Exothermic Processes

On a sunny winter day, the snow on a rooftop begins to melt. As the melted water drips from the roof, it refreezes into

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icles. Describe the direction of heat flow as the water freezes.

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- Thermochemistry –The study of the transfer of heat (thermal energy) in chemical reactions. –Exothermic - transfer of heat from the system to the surroundings
 $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{energy}$
- Endothermic - the transfer of heat from the surroundings to the system energy + ...

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1. Consider 2 metals, A and B, each having a mass of 100 g and an initial temperature of 20 °C. The specific heat of A is larger than that of B. Under the same heating conditions, which metal would take longer to reach 21 °C? Explain your reasoning. 2.

Thermochemistry Worksheet #1 1. The reaction of magnesium with sulfuric acid was carried out in a calorimeter. This

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reaction caused the temperature of 27.0 grams of liquid water, within the calorimeter, to raise from 25.0 C to 76.0 C. Calculate the energy associated with this reaction. 2.

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 $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{energy}$

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–Endothermic - the transfer of heat from the surroundings to the system energy + ...

Thermochemistry: The Heat of Neutralization: Lab Report Name: _____
Partner(s): _____ Section Number: _____
Lab Report - Turn in pages 4 – 7 for your lab report. DATA Record time and temperature data for all three reactions in the table below until the temperatures ...

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